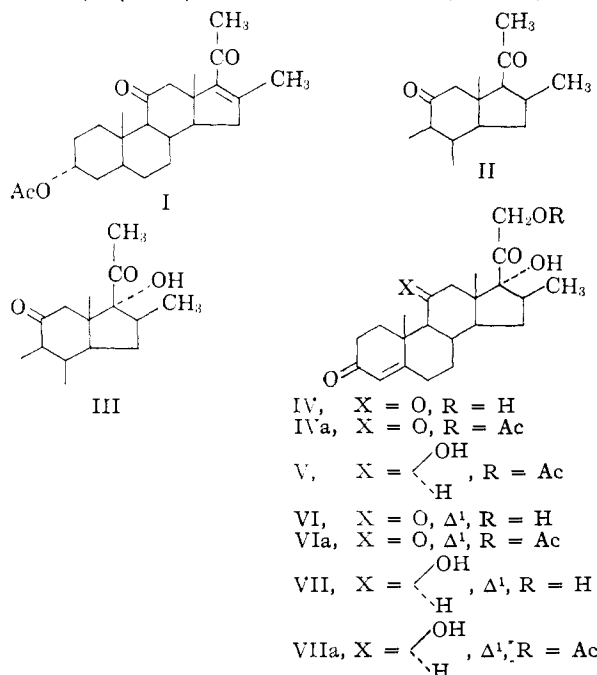


16 β -METHYL CORTICAL STEROIDS

Sir:

We wish to report the preparation of 16 β -methyl homologs of cortisone and its congeners. These substances are the first β -substituted cortical steroid derivatives to be reported which are more potent anti-inflammatory agents than the corresponding parent steroids. Previously described potentiating substituents have been attached to the α -face of the steroid nucleus.¹

Addition of diazomethane² to 3 α -acetoxy-16-pregnene-11,20-dione³ gave the corresponding pyrazoline derivative, m.p. 186–190° dec., *Anal.* Found: C, 69.37; H, 8.01, which was converted by pyrolysis to 3 α -acetoxy-16-methyl-16-pregnene-11,20-dione (I), m.p. 165–167°, $[\alpha]_D^{25}$ +75°, λ_{max}^{MeOH} 249 m μ (9,300). *Anal.* Found: C, 74.30; H,



8.80. Hydrogenation of I in methanol over palladium on calcium carbonate led smoothly to 3 α -acetoxy-16 β -methylpregnane-11,20-dione (II), m.p. 160–163°; $[\alpha]_D^{25}$ +96°. *Anal.* Found: C, 74.15; H, 9.15. Reaction of II with acetic anhydride-perchloric acid⁴ in chloroform-carbon tetrachloride led to the corresponding 17(20)-enol acetate, which with perbenzoic acid⁵ in benzene followed by alkaline hydrolysis afforded 3 α ,17 α -dihydroxy-16 β -methylpregnane-11,20-dione (III), m.p. 192–197°, $[\alpha]_D^{25}$ +67°. *Anal.* Found: C, 72.97; H, 9.25. Confirmation of the structure of III was obtained by acetylation of III at C-3 followed by cleavage oxidation⁶ to the known 3 α -ace-

toxy-16 β -methylpregnane-11,17-dione.⁷ Bromination of III yielded the corresponding 21-bromide, m.p. 165–175° dec. *Anal.* Found: C, 59.34; H, 7.55; Br, 18.14, which on treatment with potassium iodide and potassium acetate in refluxing acetone gave 3 α ,17 α ,21-trihydroxy-16 β -methylpregnane-11,20-dione 21-acetate, m.p. 222–228°, $[\alpha]_D^{25}$ +124.5°. *Anal.* Found: C, 68.79; H, 8.44. Oxidation of the latter compound with N-bromosuccinimide in aqueous *t*-butanol afforded 17 α ,21-dihydroxy-16 β -methylpregnane-3,11,20-trione 21-acetate, m.p. 210–212°, $[\alpha]_D^{25}$ +130°. *Anal.* Found: C, 68.81; H, 7.91, which on monobromination at C-4 followed by dehydrobromination *via* the 3-semicarbazone produced 16 β -methylcortisone acetate (IVa), m.p. 230–236°, $[\alpha]_D^{25}$ +252°, λ_{max}^{MeOH} 238 m μ (15,800). *Anal.* Found: C, 69.24; H, 7.58. Hydrolysis of IVa by means of potassium bicarbonate in aqueous methanol yielded 16 β -methylcortisone (IV), m.p. 205–210°, $[\alpha]_D^{25}$ +237°, λ_{max}^{MeOH} 238 m μ (16,200). *Anal.* Found: C, 70.84; H, 8.40. IV was converted *via* its 3,20-disemicarbazone⁸ by hydride reduction, reversal and acetylation to 16 β -methylhydrocortisone acetate (V), m.p. 220–225°; $[\alpha]_D^{25}$ +187°, λ_{max}^{MeOH} 242 m μ (15,800). *Anal.* Found: C, 69.10; H, 7.89.

Dehydrobromination of the 2,4-dibromide of 17 α ,21-dihydroxy-16 β -methylpregnane-3,11,20-trione 21-acetate led to 16 β -methylprednisone acetate (VIa), m.p. 230–233°, $[\alpha]_D^{25}$ +216°, λ_{max}^{MeOH} 238 m μ (15,100). *Anal.* Found: C, 69.25; H, 7.25. Potassium bicarbonate-aqueous methanol hydrolysis of VIa afforded 16 β -methylprednisone (VI), m.p. 195–200°, $[\alpha]_D^{25}$ +205°, λ_{max}^{MeOH} 238 m μ (14,900). *Anal.* Found: C, 70.91; H, 7.57. 16 β -Methylhydrocortisone 21-acetate (Va) was dehydrogenated by reaction with selenium dioxide⁹ to 16 β -methylprednisolone acetate VIIa, m.p. 209–214°, λ_{max}^{MeOH} 243 m μ (15,000). *Anal.* Found: C, 68.76; H, 7.55, which on treatment with potassium bicarbonate in aqueous methanol was converted to 16 β -methylprednisolone VII, m.p. 205–210°, $[\alpha]_D^{25}$ +145°, λ_{max}^{MeOH} 243 m μ (14,600). *Anal.* Found: C, 70.57, H, 8.08.

16 β -Methyl-9 α -fluoroprednisolone acetate, m.p. 205–208°, 16 β -methyl-9 α -fluoroprednisolone, m.p. 225–227°, and related compounds have been prepared and will be described in a forthcoming publication.

This series of compounds exhibited unusually high anti-inflammatory activity comparable to that observed with the corresponding 16 α -methyl corticosteroids.¹ Details of the biological findings will appear elsewhere.

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(1) Cf. G. E. Arth, D. B. R. Johnston, J. Fried, W. W. Spooncer, D. R. Hoff, L. H. Sarett, *This Journal*, **80**, 3160 (1958); G. E. Arth, J. Fried, D. B. R. Johnston, D. R. Hoff, L. H. Sarett, R. H. Silber, H. C. Stoerk and C. A. Winter, *ibid.*, **80**, 3161 (1958), and references cited therein.

(2) Cf. A. Wettstein, *Helv. Chim. Acta*, **27**, 1803 (1944).

(3) H. L. Slates and N. L. Wendler, *J. Org. Chem.*, **22**, 498 (1957).

(4) Cf. D. H. R. Barton, R. M. Evans, J. C. Hamlett, P. G. Jones and T. Walker, *J. Chem. Soc.*, 747 (1954).

(5) Cf. T. H. Kritchevsky, D. L. Garmaise and T. F. Gallagher, *This Journal*, **74**, 483 (1952).

(6) Cf. N. L. Wendler, D. Taub and H. L. Slates, *ibid.*, **77**, 3559 (1955).

(7) N. L. Wendler and D. Taub, *Chemistry and Industry*, 415 (1958).

(8) Cf. N. L. Wendler, Huang-Minlon and M. Tishler, *This Journal*, **73**, 3818 (1951). See also R. E. Jones and S. A. Robinson, *J. Org. Chem.*, **21**, 586 (1956).

(9) Cf. Ch. Meystre, H. Frey, W. Voser and A. Wettstein, *Helv. Chim. Acta*, **39**, 734 (1956).